

In-Vitro Antioxidant Activity, Acute Oral Toxicity Studies and Preliminary Phytochemical Characterization of the Bark Extract of *Terminalia arjuna* (L.)

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Abstract: The free radicals and the reactive oxygen species (ROS) are known to induce oxidative stress and it has been implicated in the pathology of cardiovascular diseases, inflammatory conditions, cancer and ageing. The activities associated with ROS can be delayed, prevented or removed by antioxidant compounds (natural or synthetic). The use of synthetic antioxidants restricted because of their known side effects such as liver damage and carcinogenesis. The aim of this study to evaluate *in vitro* antioxidant and acute oral toxicity of *Terminalia arjuna* extracts. The present finding reveals that the purified fraction at 100 µg/ml, showed maximum (91.32 ± 0.10 %,) DPPH radical scavenging effect in comparison with standard ascorbic acid (79.46 ± 0.10%) at the 10 µg/ml concentration. The reducing power of the purified extract was found to be dose dependent. Food and water intake of the animals in test and control groups was found normal during the 14 day acute oral toxicity studies and no apparent changes were observed in the internal organs of both, the test and control groups, after gross necropsy. The preliminary phytochemical screening of the crude acetone extract revealed dominant presence steroids, terpenoids, polyphenols, alkaloids and tannins. TLC profile of the purified fraction revealed a single band of R_f 0.38, a characteristic feature of triterpenoids. The UV absorption maximum of the purified fraction was recorded at 194nm. The FT-IR spectrum indicated presence of aromatic rings 3421 (COOH), 2957 (alkanes, CH₂ and CH₃), 1726 (carbonyl), 1599 (carboxylic acid), and region between 1000-1300 stretching of C-O, ester and ether carboxylic group. Thus, the isolated bioactive phytoconstituents form the bark extract of *Terminalia arjuna* could be used as natural anti-oxidants.

Keywords: Antioxidant activity, Acute oral toxicity, *T. arjuna*, TLC, Triterpenoids.

INTRODUCTION

Free radicals and reactive oxygen species (ROS) generated in the biological system are the major cause of the degenerative conditions such as aging, cancer, inflammation and atherosclerosis [1]. Hydroxyl radical (OH[•]), superoxide anion (O₂^{•-}) and nitric oxide (NO[•]) are known to cause membrane damage, protein denaturation and lipid peroxidation. Natural antioxidant enzymes present in the body as well as dietary intake of antioxidants circumvents free radicals and reactive oxygen species generated in the body. Antioxidants also have the ability to prevent the oxidative stress generated by the reactive oxygen species and consequently the free radical mediated oxidative damage in the cell [2]. Sufficient amounts of exogenous antioxidants are required to reduce the damage caused by free radicals and reactive oxygen species. Therefore, there is great need to search for safe, less

cytotoxic and chemo-preventive natural antioxidants. Antioxidant potential of medicinal plants attributed by the bio-active compounds present in it. Medicinal plants are rich source of diverse secondary metabolites having potential to reduce reactive oxygen species and thus widely recognized for their pharmaceutical and medicinal importance. A number of clinical studies have reported that antioxidants of plant origin such as polyphenols, flavonoids, terpenoids, tannins etc. are reported to reduce the oxidative stress in the biological system [3].

Indian traditional system of medicine has identified a large numbers of plants for their antioxidative potential and human diseases. Medicinal plants such as *Celastrus peniculatus*, *Carrisa carrandus*, *Achyrahes aspera*, *Cassia auriculata*, *Coccinia indica*, *Mentha spicata*, *Hygrophilla auriculata*, *Datura stramonium*, *Delonix regia*, *Coriandrum sativum* and *Pterospermum acerifolium* are known to have antioxidant activities [4]. *Terminalia arjuna* (Combretaceae), commonly known as Arjun tree, is a large deciduous tree known for its cardioprotective role. The bark of *T. arjuna* has been recommended and used as a cardiac tonic and bark

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powder/decoction is used to treat heart diseases, bone fractures, skin diseases, poly-urea, white discharge, giddiness, fever and worms.

The phytochemistry of *T. arjuna* is well understood. The bark powder is known to contain tannins, triterpenoids (arjunic acid and arjungenin) and terpenoids glycosides like arjunnetin and arjunglucoside [5]. Several experimental and clinical studies have been undertaken and proved the beneficial effect of the *T.arjuna* bark in ischemic heart disease and cardiovascular complications [6]. However, safety and the mechanism of action of *arjuna* phytochemicals is uncertain. The present study demonstrates the *in vitro* antioxidant activity as well as evaluates the safety and toxicity by studying the acute oral toxicity of various extracts of *T.arjuna*. The study also aims at isolation and partial characterization of the bioactive fraction using chromatographic and spectroscopic techniques.

MATERIALS AND METHODS

Plant Material and Preparation of Extract

The bark of plant *T. arjuna* was collected from Toranmal Hill Station (MS, India). Plant material was identified and authenticated by the expert taxonomist and the voucher specimen (RCP-09/2014) was deposited in Department of Botany R. C. Patel Arts, Science College Shirpur. The bark collected from healthy plant was cut to obtain thin slices and subjected for shade drying at room temperature. The shade dried bark ground in a mechanical grinder to obtain a fine powder. The bark powder was further subjected to successive extraction in the Soxhlet extractor using solvents with increasing polarity. The organic phase of the extract was collected using separating funnel to obtain crude extract. The extract from separating funnel was filtered using Whatman filter paper No. 1 and subjected to dryness in rotary vacuums evaporator (Equitron, India) at high pressure and stored in refrigerator until used.

Antioxidant Activity Using DPPH Radical Scavenging Assay

The antioxidant activity was evaluated by means of measuring the 1,1-diphenyl 2-picrylhydrazyl (DPPH) radical scavenging activity of the test extracts [8]. The method is based on reduction of DPPH in the presence of hydrogen-donating antioxidant due to the formation of the non-radical form of DPPH-H. This conversion indicated by color change from purple to yellow, it was

measured spectrophotometrically at 517nm [10]. Briefly, freshly prepared 1 ml of methanolic DPPH (2 mM) was mixed with different concentration of extract and ascorbic acid standard (Vitamin C), The mixture was shaken vigorously and incubate for 30 min in the dark, and the absorbance was measured at 517 nm against a blank. The percent DPPH radical scavenging ability was calculated using the following formula.

$$\text{DPPH radical scavenging ability (\%)} = (A_0 - A_1 / A_0) \times 100$$

Where, A_0 is the absorbance of control, A_1 is the absorbance of sample. All samples were analyzed in triplicate.

Reducing Power Assay

The test extracts with different concentrations were mixed with 0.2 M, 2.5 ml phosphate buffer (pH 6.6) and 1%, 2.5 ml potassium ferricyanide. The mixture was incubated for 20 min at 50°C and cooled, and then 10%, 2.5 ml tricarboxylic acid was added and the mixture was centrifuged for 10 min at 600 g. The upper layer of the solution (2.5 ml) was mixed with 2.5 ml of distilled water and 0.1%, 0.5 ml ferric chloride (FeCl_3). The absorbance was measured at 700 nm. [8]. Ascorbic acid and phosphate buffer were used as standard and blank, respectively. All samples were analyzed in triplicate.

Acute Oral Toxicity

The purpose of acute oral toxicity testing of the plant and their preparations influence their safety to human, particularly for a pharmaceutical preparation. Acute oral toxicity testing was performed according to the OECD guideline no 423 as per the CPCSEA approval (SPTM-IAEC/Dec 14/7/18). Young, healthy 7-8 week Albino mice were employed for the testing and the animals were fasted for 3-4h before dose. A single dose of crude extract and purified fraction of *T arjuna* at 300, 1000 and 2000 mg/kg body weight was given orally. After administration of dose, the food was withheld for 1-2h. For feeding, conventional animal diet was used and animals were kept in the separate polypropylene cages. Animals were observed for clinical signs of toxicity at 0-0.5, 0.5-1, 1-2, 2-4, 24-48h post dosing with special attention during first 4 h. If no mortality was observed for 48 h, other mice were administered with dose of 1000 mg/kg body weight and observed for same symptoms. In the absence of toxicity, mice were administered with dose 2000 mg/kg body weight and checked for mortality or toxicity. Cage

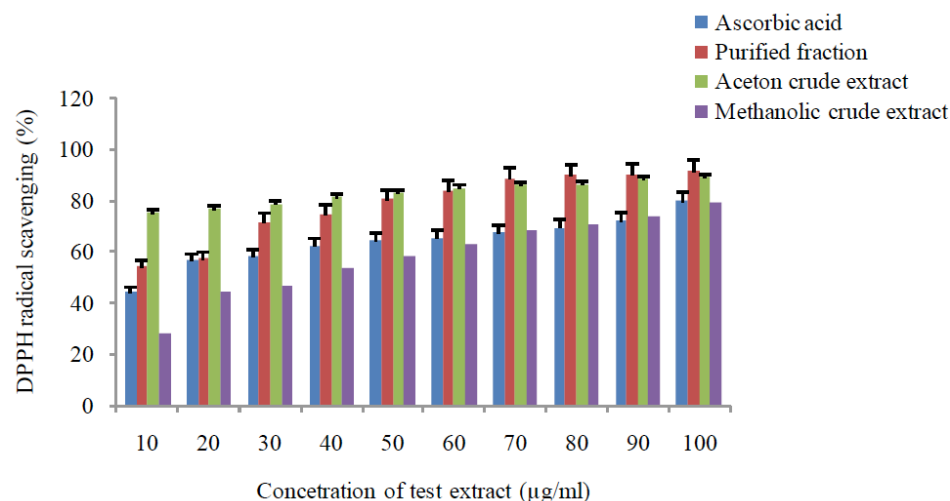


Figure 1: % DPPH radical scavenging effect of acetone, crude extract, and purified fraction and standard (ascorbic acid). Data are shown as mean \pm SD of values obtained in three separate experiments.

side observation includes changes in skin fur, eyes and mucous membrane (nasal), respiratory rate, heart rate and blood pressure, salivation, lacrimation, perspiration and CNS changes.

Isolation and Characterization of Bioactive Metabolites

The crude acetone extract after drying in vacuum afforded a brown solid which re-dissolved in methanol. The preliminary phytochemical investigation of the test extracts was carried out according to previously reported methods [9]. The resulting extract was subjected for Thin-Layer Chromatography (TLC). Five microliters of sample was applied using Spraylin-V applicator (Aetron, India) on aluminum TLC plate coated with Silica gel G_{F254} (Merck, India) and developed in ascending mode using different mobile phases. For separation of phytoconstituents, the extract was dissolved in acetone and adsorbed on silica gel of 60-120 mesh size for column chromatography. The air dried slurry was applied onto the top of column which was preconditioned with hexane. The column was eluted with hexane (100 %), hexane: ethyl acetate (80:20, 60:40 and 50:50 v/v); chloroform: methanol (70:30, 50:50, and 20:80 v/v) and finally with 100% methanol. The fractions were collected separately and matched by TLC fingerprinting for tentative identification and homogeneity.

UV and FT-IR Spectral Analysis

The purified fraction obtained from silica gel column was subjected for UV and FT-IR spectroscopic characterization. The UV spectrum of purified fraction eluted from the column was recorded between 200-700

nm by UV-Visible spectrophotometer (Shimadzu 1750, Japan). FT-IR spectrum of the dried powder was recorded by FT-IR spectrophotometer (Shimadzu 8202 PC, Japan 4000 – 400 cm^{-1}) using KBr as a matrix.

RESULTS

Antioxidant Activity by DPPH Radical Scavenging Assay

The DPPH radical scavenging activity of the acetone crude extract, purified fraction and ascorbic acid was found to be dose dependent and the results are shown in Figure 1. The crude acetone extracts of *T. arjuna* showed $88.80 \pm 0.10\%$ scavenging effect. The results showed that, the purified fraction at 100 $\mu\text{g/ml}$, showed maximum ($91.32 \pm 0.10\%$) scavenging effect in comparison with standard ascorbic acid ($79.46 \pm 0.10\%$). The IC_{50} ($\mu\text{g/ml}$) of the crude acetone extract and purified fraction of *T. arjuna* was found to be 140 ± 11.21 and 45.95 ± 5.74 , respectively (Table 1).

Table 1: IC_{50} of DPPH Radical Scavenging Activity

Activity	IC_{50} ($\mu\text{g/ml}$)
Purified fraction	45.95 ± 5.74
Acetone crude extract	140 ± 11.21
Methanol crude extract	79.37 ± 8.55
Ascorbic acid	6.9 ± 0.4

Reducing Power Assay

The results of reducing power of the acetone and methanol crude extracts, purified fraction and ascorbic

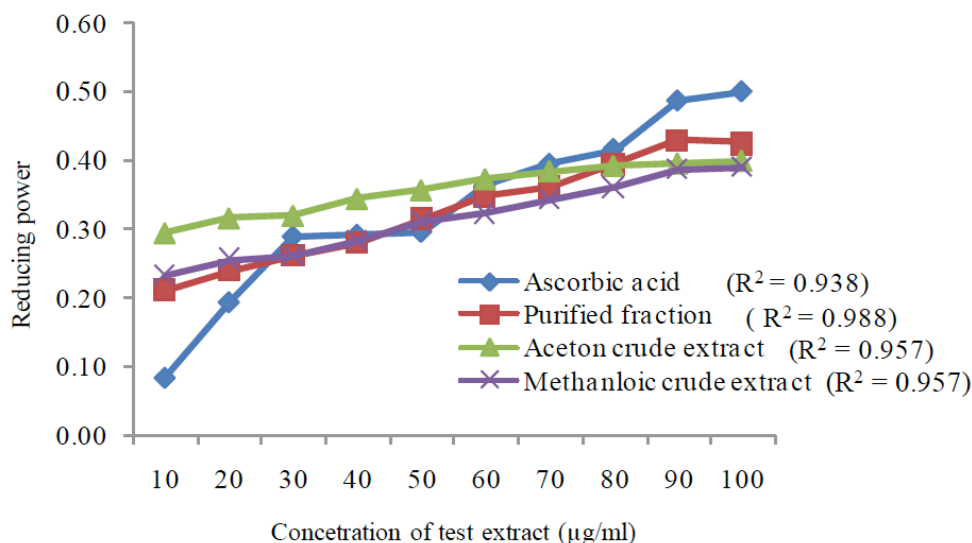


Figure 2: Reducing power of acetone crude, methanol crude, purified fraction and ascorbic acid data shown as mean \pm SD.

acid standard are shown in Figure 2. Reducing power of the purified fraction was found to be dose dependent ($R^2 = 0.988$). The results indicate that reducing potential of the purified fraction is significantly higher than that of acetone; methanol crude extract and is comparable with the ascorbic acid standard. The results clearly demonstrate the admirable reducing power potential of purified fraction. The bioactive constituents present in the extracts may have potential donating activities which lead to their strong reducing potential.

Acute Oral Toxicity

The results of acute oral toxicity studies are summarized in Tables 2 and 3. The food and water intake of the animals in test and control groups was found normal (Table 3). During the 14 day oral acute toxicity testing, no mortality, no sign of toxicity and negative symptoms was observed in any animals treated with purified fraction of the *T.arjuna*. There is no statistically significant difference ($P < 0.05$) was observed in water and food intake between vehicle

control and test group (Table 3). The changes in body weight at 300, 1000, and 2000 mg/kg body weight of *T.arjuna* purified fraction was not significant (Figure 3). The results of acute treatment effect of the *T. arjuna* purified fraction on organ weight in mice are shown in Table 2. No apparent changes were observed in the organs of both, the test and control groups, after gross necropsy.

Chromatographic and Spectroscopic Characterization of the Active Principle

The preliminary phytochemical screening of the crude acetone extract revealed dominant presence of phytoconstituents such as steroids and terpenoids, polyphenols, alkaloids and tannins. The crude extract of *T. arjuna* was subjected for column chromatography using optimized solvent system (non-polar to polar mobile phases) to elute out bioactive components. The column eluted with chloroform: ethyl acetate (1:1) and a total of 100 fractions (20 ml each) were collected. The fraction number thirty to sixty affording semisolid colorless compound does not showed significant

Table 2: Acute Treatment Effect of *T.arjuna* Purified Fraction on Organ Weight

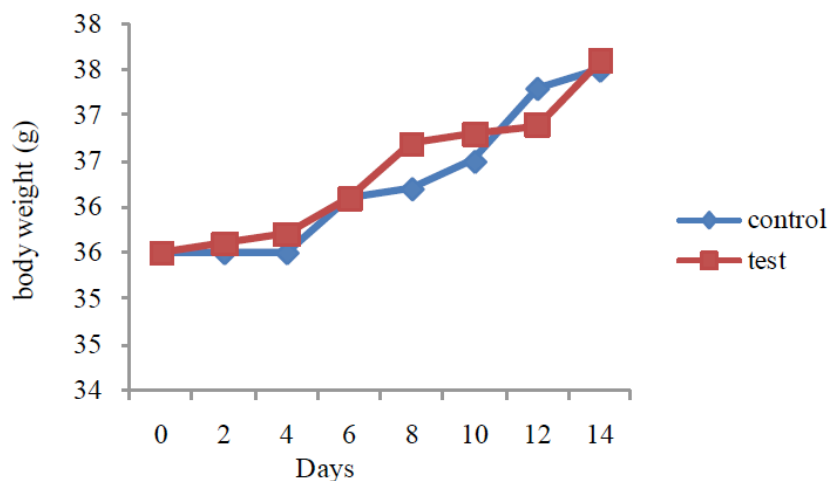
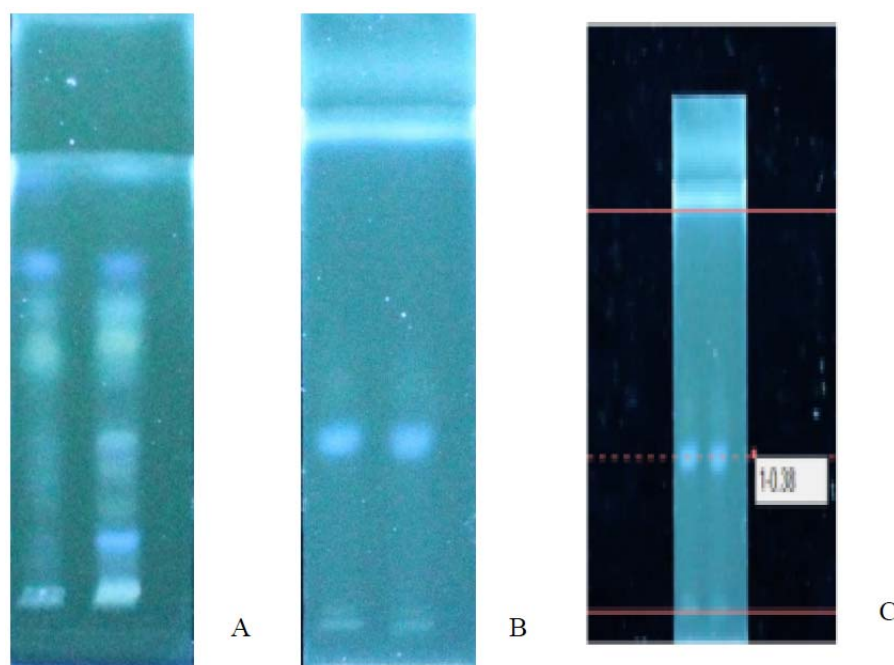
Organ	Weight in grams	
	Control	<i>T. arjuna</i> purified fraction
Lungs	1.89 \pm 0.11	1.84 \pm 0.20
Liver	1.42 \pm 0.25	1.29 \pm 0.26
Spleen	0.19 \pm 0.28	0.15 \pm 0.24
Heart	0.14 \pm 0.15	0.17 \pm 0.10
Kidney	0.43 \pm 0.26	0.41 \pm 0.35

Data expressed mean \pm SD.

Table 3: Food and Water Intake During Acute Toxicity Study with Purified Fraction

Parameters	Control	<i>T. arjuna</i> purified fraction
Water (mL/day)	6.98±0.57	7.15±0.48 ^a
Food (g/day)	2.91±0.35	3.25±0.27 ^a

Data are expressed as mean ±SD. *P* < 0.05, are considered for statistically significant between groups.

**Figure 3:** Changes in the body weight after administration *T. arjuna* purified fraction.**Figure 4:** TLC profile of *T. arjuna* (A) Crude extract (B) Purified fraction, *R_f* of purified fraction (C).

bioactivity (data not shown) and hence discarded. Fraction number sixty five to hundred afforded a colorless solid showing single band of *R_f* 0.38 (Figure 4C) in different mobile phases showed excellent bioactivity hence subjected to further characterization using spectroscopic and chromatographic techniques.

Among the various mobile phases tried for TLC separation, better separation was obtained with chloroform: ethyl acetate (1:1v/v) when 5 % sulfuric acid was used as spraying reagent (Figure 4A). The TLC separated band of purified fraction showed fluorescence at longer wavelength (Figure 4B) which is

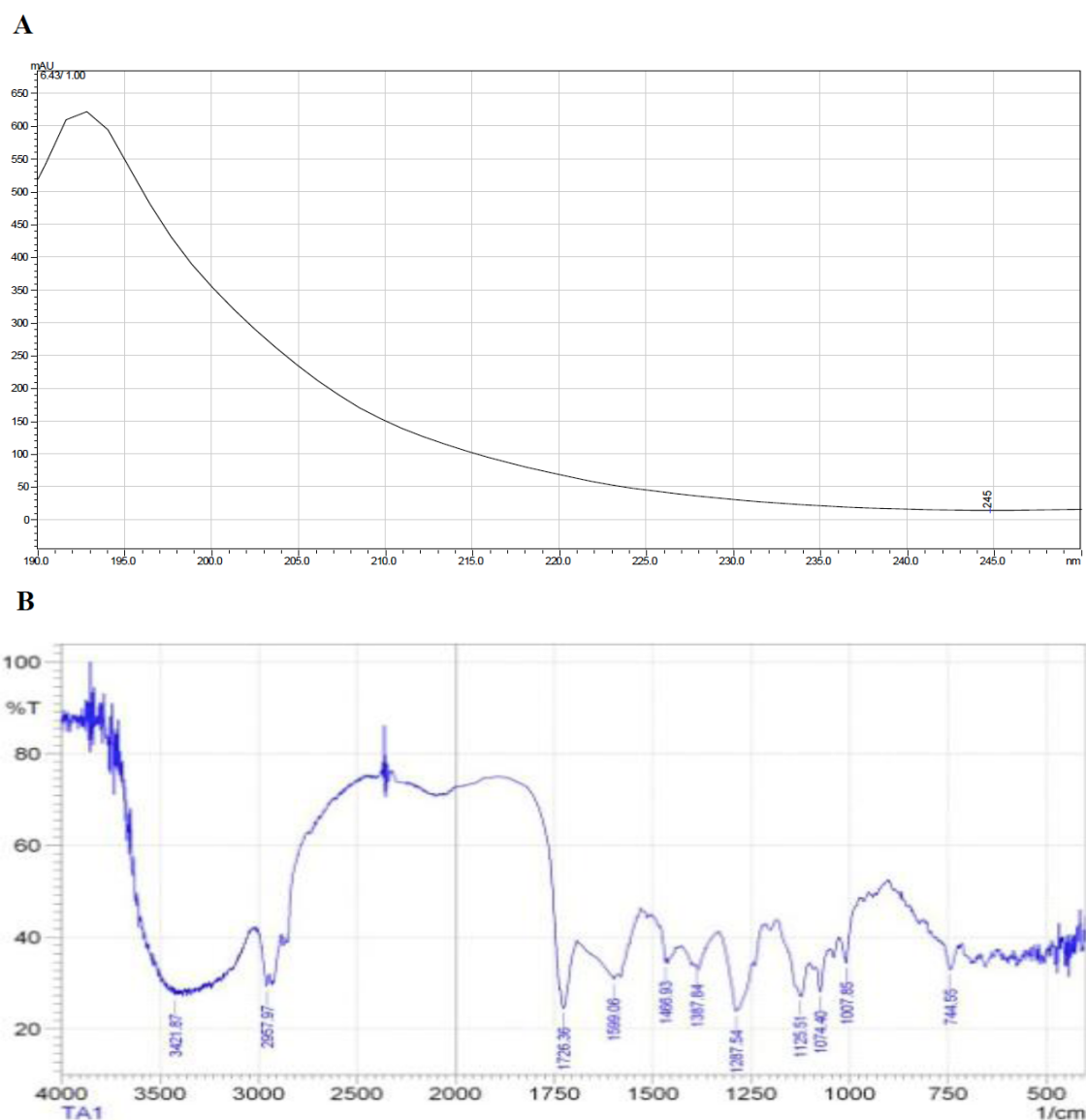


Figure 5: UV (A) and FT-IR (B) spectrum of the purified fraction of *T. arjuna*.

the characteristic feature of triterpenoids. UV spectrum of purified fraction of R_f 0.38 showed maximum absorption at 194 nm (Figure 5A). The FT-IR spectrum (Figure 5B) showed presence of aromatic rings. A stretching at 3421 (COOH), 2957 (alkanes, CH₂ and CH₃), 1726 (carbonyl), 1599 (carboxylic acid), and region between 1000-1300 (C-O, ester and ether carboxylic group) was observed. On the basis of TLC characteristics, UV and FT-IR spectrum, the purified fraction may contain terpenoids as a major constituent.

DISCUSSION

Natural products from plant sources have various protective and therapeutic effects and they are essential for preventing various diseases [11]. Many Indian medicinal plants such as *Borago officinalis*, *Capparis sicula* subsp. *Sicula*, *Malva sylvestris*, *Mentha*

aquatic, and *Raphanus raphanistrum* have been explored as natural source of strong antioxidants [12]. Our study demonstrated the strong antioxidant potential and acute oral toxicity of the bark extract of *T. arjuna*. Different physiological and pathological conditions generate large number of free radicals reactive oxygen species and reactive nitrogen species. Free radicals adversely affect lipids, proteins, and DNA and trigger a number of human diseases including aging and atherosclerosis. Thus a balanced intake of dietary antioxidants circumvents free radicals mediated damage. The antioxidant metabolites from *T. arjuna* in our studies are effective yet safe and thus can be effectively used as to prevent oxidative stress mediated cellular damages. Moreover, the IC₅₀ value of purified fraction (IC₅₀ = 45.95 ± 5.74 µg/ml) was found less than previous reports [12, 13].

Natural antioxidants have ability to prevent the formation of ROS by donating electron and thus they prevent the oxidative DNA damage and lipid peroxidation (Rice-Evans *et al.* 1997). Thousands of medicinal plants available that contain phytochemicals with an excellent antioxidant potential. Plant secondary metabolites like phenolics and flavonoids with antioxidant activity are widely studied. These compounds play an important role to adsorb and neutralize free radicals, and quench singlet and triplet oxygen.

However, relatively few reports are exists on terpenoids. Moreover, lack of knowledge on safety and toxicity of the plant metabolites is the major constrain for their use in humans. The purified fraction in our studies showed no sign of toxicity and its maximum tolerated dose is more than 2000 mg/kg body weight. However further repeated acute toxicity study is required for the validation of purified fraction as safer, nontoxic antioxidant. Also, *in-vivo* clinical trials are required for the proof of its strong antioxidant activity. It can be concluded that, the purified fraction of *T. arjuna* has excellent free radical scavenging activity and reducing power and it was found better than standard, ascorbic acid. The bioactive principle of the purified fraction of *T. arjuna* could be terpenoids which are safe and could be used as natural source of antioxidants for providing alternative remedy for the management of oxidative stress.

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COMPETING INTEREST

Authors declare that there are no conflicts of interest.

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