## SUPPLEMENTARY INFORMATION

### **General Procedures of Preparations**

Chemicals of the highest commercial grade available, purchased from Wako, TCI and Kanto were used as received without further purification. (2S, 3S)-3-Aminoaspartic acid was prepared according to the literature procedures (Ibarra AG *et al.* J Org Chem 1997; 62: 2478).

#### Table S1: Crystal Data and Structure Refinement for 2, 4 and 6

	2	4	6
Empirical formula	C <sub>19</sub> H <sub>13</sub> Cl <sub>4</sub> CuN <sub>2</sub> O <sub>7</sub>	$C_{35}H_{49}CuN_2O_7$	$C_{19}H_{17}CuN_2O_9$
Formula weight	586.67	673.32	480.89
Crystal system	Monoclinic	Triclinic	Monoclinic
Space group	<i>P</i> 2 <sub>1</sub> /n (#14)	<i>P</i> 1 (#1)	<i>P</i> 2 <sub>1</sub> /n (#14)
a (Å)	10.880 (5)	13.661 (10)	14.34 (4)
b (Å)	10.559 (5)	17.014 (14)	19.24 (4)
<i>c</i> (Å)	28.702 (13)	11.644 (9)	13.77 (4)
α (°)	100.429 (16)	112.63 (3)	91.3 (2)
β (°)		106.69 (6)	
γ (°)		75.46 (6)	
V (Å <sup>3</sup> )	3243 (3)	2362 (3)	3798 (17)
Z	4	1	4
Rwp (%)	19.24	26.83	17.87

## Table S2: Selected Bond Lengths and Angles for 2

Cu 1-O 1	1.9326(6)Å	O 1-Cu 1-O 2	92.862(17)°
Cu 1-O 2	1.8985(8)Å	O 1-Cu 1-N 1	91.25(2)°
Cu 1-O 3	2.2532(7)Å	O 1-Cu 1-N 2	170.671(4)°
Cu 1-N 1	1.9740(7)Å	O 2-Cu 1-O 3	91.01(2)°
Cu 1-N 2	2.0033(7)Å	O 2-Cu 1-N 1	171.695(4)°
		O 2-Cu 1-N 2	92.705(17)°
		N 1-Cu 1-N 2	84.20(2)°

#### Table S3: Selected Bond Lengths and Angles for 4

Cu 1-O 1	1.9108(13)Å	O 1-Cu 1-O 2	92.62(7)°
Cu 1-O 2	1.9128(15)Å	O 1-Cu 1-N 1	91.90(7)°
Cu 1-O 3	2.0546(16)Å	O 1-Cu 1-N 2	171.203(9)°
Cu 1-N 1	1.9720(16)Å	O 2-Cu 1-O 3	96.09(6)°
Cu 1-N 2	1.9694(13)Å	O 2-Cu 1-N 1	171.208(10)°
		O 2-Cu 1-N 2	91.90(7)°
		N 1-Cu 1-N 2	84.65(7)°

Cu 1-O 1	1.922(4)Å	O 1-Cu 1-O 2	93.9(2)°
Cu 1-O 2	1.914(4)Å	O 1-Cu 1-N 1	171.26(3)°
Cu 1-O 3	2.514(5)Å	O 1-Cu 1-N 2	91.6(2)°
Cu 1-N 1	1.984(4)Å	O 2-Cu 1-O 3	84.38(17)°
Cu 1-N 2	1.985(4)Å	O 2-Cu 1-N 1	92.14(19)°
		O 2-Cu 1-N 2	171.678(4)°
		N 1-Cu 1-N 2	83.12(19)°

## Table S4: Selected Bond Lengths and Angles for 6

# Table S5: Electron Transition of Chiral Salen Cu(II) Complexes 1-9

	π-π*	d-d
1	HOMO -2 $\rightarrow$ LUMO	HOMO -4 $\rightarrow$ LUMO
2	$HOMO \rightarrow LUMO + 1$	HOMO -5 $\rightarrow$ LUMO
3	HOMO -1 $\rightarrow$ LUMO +1	HOMO -5 $\rightarrow$ LUMO
4	HOMO $\rightarrow$ LUMO +1	$HOMO \rightarrow LUMO +2$
5	HOMO -2 $\rightarrow$ LUMO +3	HOMO -5 $\rightarrow$ LUMO
6	$HOMO \rightarrow LUMO + 1$	HOMO -4 $\rightarrow$ LUMO
7	HOMO -1 $\rightarrow$ LUMO +2	HOMO $-3 \rightarrow$ LUMO
8	$HOMO \rightarrow LUMO +2$	HOMO $-3 \rightarrow$ LUMO
9	$HOMO \rightarrow LUMO +2$	$HOMO \rightarrow LUMO +1$

## Table S6: Electrochemical Measurement of Chiral Salen Cu(II) Complexes 1-9

	Ered [V vs NHE]	Eox [V vs NHE]	HOMO [eV]	LUMO [eV]	HOMO-LUMO gap [eV]
1	-0.758	0.682	-4.46	-3.02	1.57
2	-0.798	0.858	-4.64	-2.98	1.66
3	-0.868	0.252	-4.03	-2.91	1.12
4	-0.829	0.388	-4.17	-2.95	1.22
5	-0.938	0.232	-4.01	-2.84	1.17
6	-0.758	0.632	-4.41	-3.02	1.07
7	-0.798	0.672	-4.45	-2.98	1.39
8	-0.838	0.732	-4.51	-2.94	1.61
9	-0.778	0.292	-4.07	-3.00	1.47

## Table S7: Absorption Wavelength, Energy Levels and HOMO-LUMO Gap of Chiral Salen Cu(II) Complexes 1-9

	λ/π-π* [nm]	λ calc./π-π* [nm]	E <sub>red</sub> [V vsNHE]	E <sub>ox</sub> [V vs NHE]	HOMO-LUMO gap [eV]	HOMO-LUMO gap calc. [eV]
1	346	381	-0.758	0.682	-3.02	4.216
2	390	398	-0.798	0.858	-2.98	4.266
3	421	410	-0.868	0.252	-2.91	4.059
4	389	394	-0.829	0.388	-2.95	4.014
5	416	408	-0.938	0.232	-2.84	4.153
6	348	405	-0.758	0.632	-3.02	4.182
7	526	389	-0.798	0.672	-2.98	3.890
8	375	392	-0.838	0.732	-2.94	4.053
9	430	429	-0.778	0.292	-3.00	4.270

	Voc [V]	Jsc [mAcm <sup>-2</sup> ]	FF [%]	η [×10 <sup>-2</sup> %]
1	0.301	0.071	29.0	0.6126
2	0.397	0.400	21.0	3.3486
5	0.208	0.142	24.0	0.7071
9	0.189	0.032	28.0	0.1690
N3	0.405	0.379	20.0	1.3664



Figure S1: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 2.





Figure S2: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 3.



Figure S3: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 4.



Figure S4: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 5.



Figure S5: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 6.





Figure S6: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 7.



Figure S7: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 8.



Figure S8: (a) Simulated UV-vis spectra, (b) simulated CD spectra and (c) molecular orbitals of the ground state (HOMO) and excited state (LUMO) for 9.